

Organocatalyzed highly atom economic one pot synthesis of tetrahydropyridines as antimalarials

Mridul Misra^a, Swaroop Kumar Pandey^b, Vivek Parashar Pandey^a, Jyoti Pandey^a, Renu Tripathi^b and Rama Pati Tripathi^{a*}

^aMedicinal and Process Chemistry and ^bParasitology Division, Central Drug Research Institute, India – 226001
Email: rpt.cdri@gmail.com; Fax: 091-(522)-2623405 / 2623938 / 2629504;

Abstract-A highly atom economic one pot synthesis of tetrahydropyridines was achieved by L-proline/TFA catalysed multicomponent reaction of β -keto-esters, aromatic aldehydes and anilines. The synthesized compounds were screened against *P. falciparum* in vitro and one of them showed antimalarial activity with MIC as low as 0.09 μ g/mL.

Keywords: Tetrahydropyridines; Antimalarial activity; L-proline; Organocatalysis; Schizont maturation; Trifluoroacetic acid

1. Introduction

Malaria is one of the most important infectious diseases claiming more than 2.7 million deaths and infecting up to 900 million people worldwide each year.¹⁻³ Currently, there are several drugs known for the treatment of this disease but only limited ones are safe drugs. The recent reports of emerging resistance against existing drugs^{3,4} has resulted in great efforts on research and development activities from academic institutes rather than pharmaceutical industries because of limited commercial opportunities. The piperidine subunit is a prominent pharmacophore, occurring in the variety of biologically important synthetic and natural products.⁵ This subunit has great chemotherapeutic values as numerous lead molecules with prominent biological activities are clinical candidates for several diseases.^{5c} Tetrahydropyridine (THP) derivatives are also useful against several metabolic disorders and human ailments. The prominent biological activities associated with this pharmacophore are antiparasitic, antimicrobial, anticancer and antiviral etc. Further, these are intricately involved in MAO based mechanism in Parkinson's disease⁶ and as inhibitors of farnesyl transferase⁷ and dihydroorotate dehydrogenase⁸ and also play key roles in many disease processes. Owing to their importance in bioorganic and medicinal chemistry, several methods including the stereocontrolled syntheses were developed and most of them are associated with many limitations.^{9,10} Aza Diels-Alder reaction and several of its modifications has extensively been used to synthesize nitrogen-containing six-membered ring compounds.¹¹ Towards this endeavor Bruce Pégot et al. have recently developed an elegant diastereoselective synthesis of 2-substituted-2,3-dihydro-4-pyridone derivatives using chiral ionic liquids.¹² Clarke et al.¹³ have also reported an InCl₃ catalyzed multicomponent synthesis of polysubstituted 1, 2, 5, 6-tetrahydropyridines in moderate to good yields. Organocatalysis, in general, provides the avenues towards green chemistry replacing toxic metal catalysts with degradable organic compounds as catalyst. It has several advantages in developing methods to access compounds in an environment friendly way. On the other hand, the atom economic synthesis¹⁴ offers the most economical method of preparing organic compounds with no loss of the starting material and therefore eco-friendly. Moreover, application of metal catalyst in the synthesis of biologically active compounds is sometimes unwanted as the presence of even minute amount of residual metal would lead to serious consequences. In view of the above, we were interested to develop a nonmetallic, organocatalytic, atom economic method for the syntheses of tetrahydropyridines and evaluate them for different biological activities. In the first instance, the synthesized compounds were evaluated against *P. falciparum* for their antimalarial activity. Our method consists of the reaction of aromatic aldehydes, anilines and a β -keto ester under the influence of L-proline/TFA as organocatalyst. The choice of L-Proline as organocatalyst is based on the facts that it acts as an excellent Lewis acid-base catalyst in a variety of organic reactions.¹⁵

2. Results and Discussion

2.1 Chemistry

The compounds (**1-21**) were synthesized by reacting β -keto esters (1.0 equivalent), aromatic aldehydes and anilines (2.0 equivalents each) in acetonitrile in the presence of L-proline and TFA (20 mol% each) at ambient temperature (Scheme 1).

Insert scheme 1

At first, aniline, 4-methoxy benzaldehyde and methyl acetoacetate were selected as the model substrates and reacted under different experimental variants (Table 1). We have compared L-proline/TFA with other catalysts including InCl_3 , silica-sulphuric acid (SSA), *ortho*-phosphoric acid, perchloric acid and TFA alone to set up standard reaction conditions. The effect of L-proline alone and in combination with TFA, using varying amount of this catalyst-combination in various organic solvents as well as in water were also studied. Among all the experimental variants the reaction with 20 mol% of L-proline and TFA in CH_3CN at ambient temperature (Table 1, entry 10) gave the best result with 72% yield of the required 1,2,6-triphenyl-4-phenylamino-1,2,5,6-tetrahydro-pyridine-3-carboxylic acid methyl ester (**1**).

Insert table 1

The structure of these compounds was established on the basis of their spectroscopic data and microanalysis. The IR spectrum of the compounds, in general, exhibited the absorption band at around $1656\text{--}1593\text{ cm}^{-1}$ indicating that the carbonyl group of carbmethoxy or carbethoxy substituents and olefinic bonds are in conjugation. The ESMS (mass spectra) of the compounds showed their respective $[\text{M}+\text{H}]^+$ peaks. In the $^1\text{H-NMR}$ spectrum, the proton at C-2 in the above tetrahydropyridines (**1-22**) was observed either as singlet at around δ 6.18-6.36 ppm or it appeared along with the multiplets of aromatic protons ranging from δ 6.70 to 6.19 ppm. The H-6 in the above compounds was apparent as broad singlet at around δ 5.00 ppm. The two methylene protons of H-5a and H-5b were appeared as two distinct *dd* [at around δ 2.62-2.76 ppm with coupling constant in the range of 5.0 (J_1) and 15.0 Hz (J_2) and at δ 2.73-2.91 ppm with coupling constant in the range of 2.0 (J_1) and 15.0 Hz (J_2) respectively] or as doublet [at around δ 2.91 ppm with coupling constant in the range of 5.0 (J)]. The aromatic protons pertaining to the phenyl substituent at C-2 and C-6 and anilinyli moiety at C-4 were observed as mixture of singlets, doublets and multiplets at around δ 6.26-8.64 ppm. The only exchangeable secondary amine proton (NH) attached at C- 4 appeared as bs at around δ 10.20 ppm. The methyl protons of carbmethoxy group appeared as singlet at around δ 3.95 ppm, while the methylene protons of carbethoxy group appeared as quartets or multiplet at around δ 4.23-4.51 ppm and methyl protons of carbethoxy group were observed as triplet at around δ 1.42-1.54 ppm respectively. It is appropriate to mention here that sometimes the quartet pertaining to the methylene protons of carbethoxy substituents appeared with two different chemical shifts due to diastereoisomeric nature of the products. In the $^{13}\text{C-NMR}$ spectrum, the C-5 carbon of the above tetrahydropyridines (**1-22**) was observed at around δ 34.0-37.6 ppm while the C-2 carbon was appeared at around δ 53.0-59.4 ppm. The C-6 and C-3 carbons were observed at around δ 54.2-61.8 and δ 96.9-102.3 ppm respectively. The aromatic carbons (ArCH) were appeared at around δ 113.4-132.3 ppm while the C-4 and quaternary aromatic carbon were observed in the range of δ 120.7-164.4 ppm. The quaternary carbon of carbethoxy and carbmethoxy group were observed at around δ 168.4-172.0 ppm The methyl carbon of carbmethoxy group appeared at around δ 51.0 ppm, while the methylene and methyl carbons of carbethoxy group appeared at around δ 60.0-64.3 and δ 15.2-19.1 ppm respectively.

Insert table 2

As evident from Table 2 the aromatic aldehydes with nitro- and benzyloxy substituents did offer only moderate yield of the product, however, other aldehydes gave good yields of the required tetrahydropyridines irrespective of the substituents in benzene ring of the aldehyde. Further, 4-methoxy aniline resulted in better yields of the required tetrahydropyridines compared to either 4-chloro- and 4-bromo aniline.

The most probable reaction mechanism (Fig. 1) involves the initial formation of an imine **A** and a Knoevenagel product **C** formed through proline catalysed enamine **B** mediated reaction. Mannich like reaction of aniline with Knoevenagel adducts **C** results an intermediate enamine **D** which undergo Aza Diels-Alder cyclization with imine **A** and results the required tetrahydropyridine **E**. The proposed reaction mechanism is supported by the fact that Knoevenagel adduct (**23**) and imine (**24**) have been isolated during one of such representative reaction of 4-methoxybenzaldehyde, methyl acetoacetate and aniline as shown in Scheme 2.

Insert Scheme 2

Insert Figure 1

The relative stereochemistry in the above compound with 2- and 6-substituents being *trans*- to each other was based on the extensive $^1\text{H NMR}$ experiments involving NOESY and NOE experiments. Absence of any NOE between H-2 and H-6 of tetrahydropyridine indicates that they are *trans* to each other, Fig. 2. Further, the $^1\text{H NMR}$ and $^{13}\text{C NMR}$ spectra of our compounds were identical to those reported earlier by Clarke et.al.¹³

Insert Figure 2

2.2 Biology

2.2.1 Results and Discussion:

The above compounds **1-21** were assayed for their blood schizontocidal activity against *P. falciparum* 3D7 strain as per earlier reported protocols.¹⁶⁻¹⁸ Initially these compounds were tested at 10.0, 5.0, 2.5 and 1.25 µg/ml conc. Compounds **1, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 15, 16, 18, 20** and **21** displayed 100% schizont inhibition at 10 µg/mL, while compounds **2, 11, 17** displayed only 33%, 23% and 83% schizont inhibition at 10 µg/mL (Table 3). Compounds **1, 3, 4, 8, 12, 16, 18** and **20** exhibited 100% schizontocidal activity at 1.25 µg/ml. These compounds were further screened for their minimum inhibitory concentration (MIC) at lower dilutions ranging from 0.78 µg/mL to 0.05 µg/mL. The only compound **4** showed 100% inhibition at 0.09 µg/ml conc., while the compounds **3, 16** and **20** showed the 100% schizont inhibition at 0.39 µg/mL concentrations. However, the compounds **1, 8, 12** and **18** exhibited slight inhibition at 0.39 µg/mL concentrations (Table 4).

The effects of various substituents in the phenyl group attached to ring carbons and C-4 nitrogen atom in tetrahydropyridines was examined in order to find out the structure-activity relationship in this series of compounds. As evident from Table 3, the introduction of fluoro- and bromo- substituents at the 4-position and chloro- at the 3-position on to the 2- and 6- phenyl substituents of tetrahydropyridines (compounds **3, 5, 13**) did not alter the schizontocidal potential whereas the only compound (**3**) exhibited a moderate enhancement in antimalarial efficacy as compared to compound **1**. However, introduction of *p*-methoxy group in the *N*- phenyl group (compounds **7, 4, 21** and **6**) resulted in marked increase in blood schizontocidal activity as compared to the tetrahydropyridines with unsubstituted *N*-phenyl moiety. In contrast to these results, the incorporation of chloro- and bromo- substituents at the 4-position of the *N*-phenyl substituent (compounds **2, 8** and **9**) resulted in loss of activity as compared to compounds **1** and **5** with unsubstituted *N*- phenyl moiety. The introduction of a benzyloxy substituent at the 4-position of the 2- and 6- phenyl substituents (compound **11**) resulted in loss of antimalarial potency as compared to compound **1**. While tetrahydropyridines with thiophenyl and pyridyl groups (compounds **15, 16, 20**) as 2- and 6- substituents resulted in enhancement in schizontocidal activity. Further, replacement of carbomethoxy substituent with carboxy group as ester moiety at C-3 in tetrahydropyridines (compounds **14, 15, 16, 17, 18, 19** and **20**) resulted in slight decrease in the blood schizontocidal activity indicating that ester group with smaller alkyl moiety is preferable over larger one.

Insert tables 3 and 4

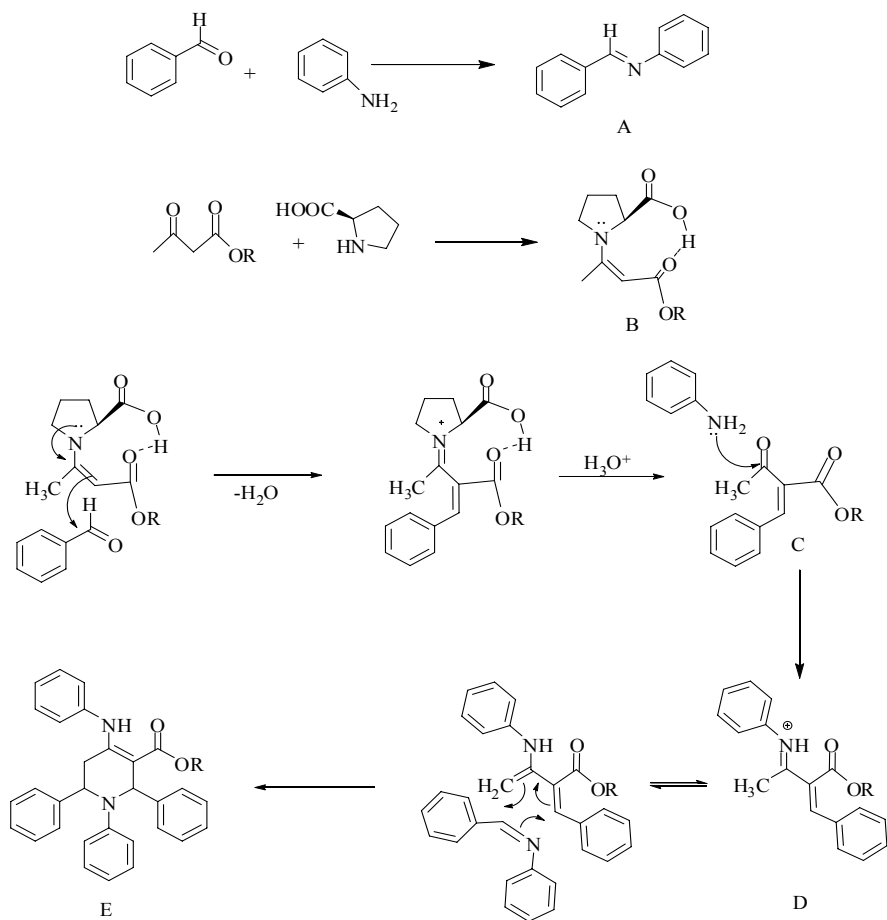
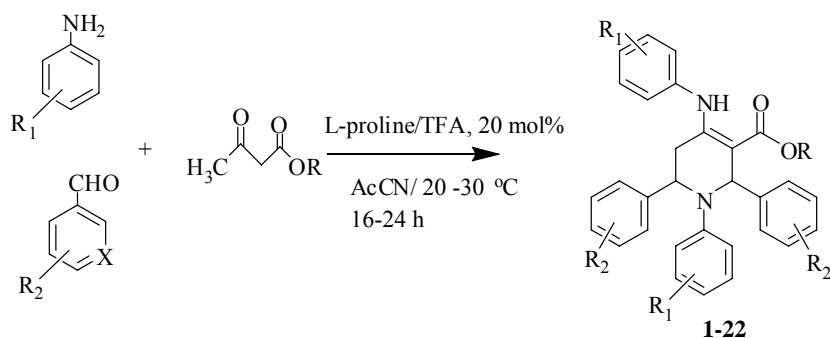
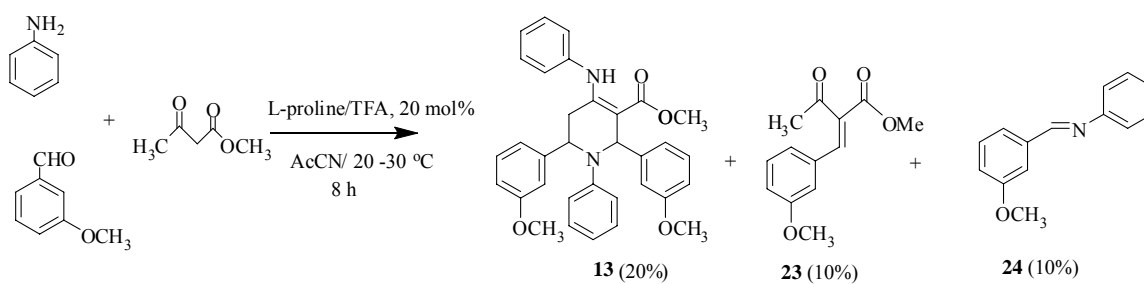


Figure 1. Proposed Reaction mechanism for the formation of tetrahydropyridines



Scheme 1. Synthesis of tetrahydropyridines from different aromatic aldehydes, anilines and β - keto esters



Scheme 2. Synthesis of tetrahydropyridine (**13**) and reaction intermediates after 8h

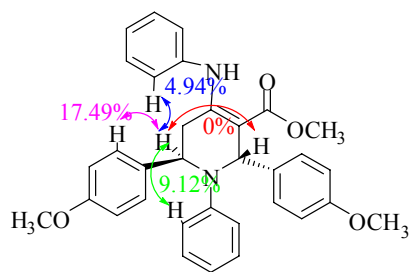


Figure 2. NOE percentage between H-2, H-6 and aromatic protons for compound 1

Table 1. Optimization experiments during reaction of 4-methoxy benzaldehyde, aniline and methyl acetoacetate

Entry	Catalyst	Solvent	Reaction time	% yields
1	InCl ₃	CH ₃ CN	50	25
2	Silica sulphuric acid	CH ₃ CN	36	10
3	<i>Ortho</i> -phosphoric acid	CH ₃ CN	24	5
4	DBU	CH ₃ CN	24	Undesired products*

5	Glycosylamino acid	CH ₃ CN	24	Undesired products
6	L- proline	CH ₃ CN	24	Undesired products*
7	L- proline + TFA (5 mol% each)	CH ₃ CN	24	45
8	L- proline + TFA (10 mol% each)	CH ₃ CN	24	50
9	L- proline + TFA (40 mol% each)	CH ₃ CN	21	72
10	L- proline + TFA (20 mol% each)	CH ₃ CN	21	72
11	-do-	THF	24	30
12	-do-	DMF	24	45
13	-do-	DMSO	24	40
14	-do-	H ₂ O	24	10

* Among the undesired product the major product was found to be Schiff's base of aldehyde and aniline

Table 2. Synthesis of tetrahydropyridines with different aromatic aldehydes, anilines and β -keto ester

Entry	Aldehyde	Aniline	β -keto compound	Product	Time (h)	Isolated Yield (%)
1	4-Methoxybenzaldehyde	Aniline	Methyl acetoacetate	1	18	70
2	4-Bromobenzaldehyde	4-Chloroaniline	Methyl acetoacetate	2	18	70
3	4-Fluorobenzaldehyde	Aniline	Methyl acetoacetate	3	24	65
4	4-Fluorobenzaldehyde	<i>p</i> -Anisidine	Methyl acetoacetate	4	18	60
5	4-Bromobenzaldehyde	Aniline	Methyl acetoacetate	5	20	65
6	3-Chlorobenzaldehyde	<i>p</i> -Anisidine	Methyl acetoacetate	6	18	60
7	4-Methoxybenzaldehyde	<i>p</i> -Anisidine	Methyl acetoacetate	7	19	75
8	4-Methoxybenzaldehyde	4-Chloroaniline	Methyl acetoacetate	8	20	70
9	4-Methoxybenzaldehyde	4-Bromoaniline	Methyl acetoacetate	9	24	65
10	4-Fluorobenzaldehyde	4-Chloroaniline	Methyl acetoacetate	10	21	65
11	4-Benzyloxybenzaldehyde	Aniline	Methyl acetoacetate	11	16	55
12	Benzaldehyde	Aniline	Methyl acetoacetate	12	17	70
13	3-Chlorobenzaldehyde	Aniline	Methyl acetoacetate	13	16	60
14	3-Chlorobenzaldehyde	<i>p</i> -Anisidine	Ethyl acetoacetate	14	22	70
15	Thiophene-2-carboxaldehyde	4-Chloroaniline	Ethyl acetoacetate	15	21	65
16	Thiophene-2-carboxaldehyde	<i>p</i> -Anisidine	Methyl acetoacetate	16	22	60
17	3-Chlorobenzaldehyde	4-Chloroaniline	Ethyl acetoacetate	17	24	65
18	Benzaldehyde	<i>p</i> -Anisidine	Ethyl acetoacetate	18	18	60
19	Benzaldehyde	4-Chloroaniline	Ethyl acetoacetate	19	22	75
20	Pyridine-3-carboxaldehyde	Aniline	Ethyl acetoacetate	20	20	55
21	4-Bromobenzaldehyde	<i>p</i> -Anisidine	Methyl acetoacetate	21	22	65
22	3-Nitrobenzaldehyde	Aniline	Methyl	22	28	70

acetoacetate

Table 3. Antimalarial activity of tetrahydropyridines against *P. falciparum* 3D7

Entry	Compound no.	% Inhibition of Schizonts of <i>P. falciparum</i> at different concentrations			
		10µg/ml	5µg/ml	2.5µg/ml	1.25 µg/ml
1	1	100	100	100	100
2	2	33	00	00	00
3	3	100	100	100	100
4	4	100	100	100	100
5	5	100	100	85	46
6	6	100	100	100	33
7	7	100	100	90	86
8	8	100	100	100	100
9	9	100	100	91	25
10	10	100	100	100	93
11	11	23	02	02	00
12	12	100	100	100	100
13	13	100	81	81	46
14	14	100	92	50	16
15	15	100	100	93	86
16	16	100	100	100	100
17	17	83	66	00	00
18	18	100	100	100	100
19	19	100	100	92	25
20	20	100	100	100	100
21	21	100	100	95	33
22	Chloroquine*	100	100	97	86
23	Control	38 % Schizonts development			

*The concentration used are 100, 50, 25 and 12.5 ng/mL.

Table 4. In vitro antimalarial activity of selected compounds (0.78-0.05 µg/mL) against *P. falciparum* 3D7

Entry	Compd. no.	% Inhibition of Schizonts at different concentrations (µg/mL) of test compounds				
		0.78	0.39	0.19	0.09	0.05
1	1	94	67	47	45	25
2	3	100	100	75	33	16
3	4	100	100	100	100	91
4	8	95	75	41	33	16
5	12	96	90	33	00	00
6	16	100	100	75	25	16
7	18	66	33	00	00	00
8	20	100	100	97	42	42
9	Chloroquine*	100	100	97	86	54
10	Control	46% Schizonts development				

*The concentration used are 100, 50, 25, 12.5 and 6.25ng/mL.

3. Conclusion

We have developed a metal free, highly atom economic organocatalytic multicomponent synthesis of tetrahydropyridines in moderate to good yields. The compounds have great potential in development of new chemotherapeutics. In the first instance, these compounds displayed antimalarial activity and their biopotential is being investigated in great detail.

4. Experimental

4.1. Chemistry

Commercially available reagent grade chemicals were used as received. All reactions were followed by TLC on E. Merck Kieselgel 60 F254, with detection by UV light and/or spraying a 20% KMnO₄ aq soln. Column chromatography was performed on silica gel (230–400 mesh, E. Merck). IR spectra were recorded as thin films or in chloroform soln with a Perkin–Elmer Spectrum RX-1 (4000–450 cm⁻¹) spectrophotometer. ¹H and ¹³C NMR spectra were recorded on a Bruker DRX-300 in CDCl₃. Chemical shift values are reported in ppm relative to SiMe₄ as internal reference, unless otherwise stated; s (singlet), d (doublet), t (triplet), m (multiplet); J in hertz. FAB mass spectra were performed using a mass Spectrometer Jeol SX-102 and ESI mass spectra with Quattro II (Micromass). Melting points were obtained manually by capillary methods and are uncorrected. Elemental analyses were performed on a Perkin–Elmer 2400 II elemental analyzer.

4.1.1 General procedure for the preparation of compounds 1-22

To a magnetically stirred solution of β-keto ester (8.0 mmol), aromatic aldehyde (14.4 mmol,) and aniline (15.6 mmol) in acetonitrile (25.0 mL), L-proline (1.5 mmol) and trifluoroacetic acid (1.09 mmol) were added and the reaction mixture was stirred at ambient temperature. The stirring continued till the disappearance of aldehyde, the reaction mixture was filtered and the solid so obtained was washed with aqueous HCl (10%) followed by water and *n*-hexane sequentially to give colourless powder which was dried under vacuum. For elemental analyses the compound was filtered through a short column of SiO₂. However, if no precipitate appears during reaction, the reaction mixture was evaporated under reduced pressure and extracted with ethyl acetate. The organic layer was washed with aqueous HCl (10%) followed by water and dried (anhydrous Na₂SO₄), evaporated under reduced pressure to give a crude mass. The latter was purified by column chromatography over SiO₂ (60-120 mesh) in EtOAc: *n*-hexane (40:60) as eluent to give the desired tetrahydropyridines 1-22.

4.1.1. Methyl 2,6-bis(4-methoxyphenyl)-1-phenyl-4-(phenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (1): White solid, mp 180 °C; R_f = 0.5 (Ethylacetate/Hexane = 1/4); FTIR (KBr): ν 3426, 2922, 2369, 1656, 1597, 1504, 1442, 1251 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 2.76 (dd, *J* = 14.8 Hz and 1.9 Hz, 1H, H-5a), 2.87 (dd, *J* = 14.9 Hz and 5.4 Hz, 1H, H-5b), 3.80 (s, 3H, ArOMe), 3.82 (s, 3H, ArOMe), 3.95 (s, 3H, OMe), 5.10 (bs, 1H, H-6), 6.37-6.39 (m, 3H, 2ArH and H-2), 6.51-6.54 (d, *J* = 8.3 Hz, 2H, ArH), 6.58-6.63 (t, *J* = 7.1 Hz, 1H, ArH), 6.73-6.84 (m, 4H, ArH), 7.05-7.33 (m, 9H, ArH), 10.31 (bs, 1H, NH) ppm; ¹³C NMR (50MHz, CDCl₃+CCl₄): δ 34.1 (CH₂, C-5), 51.3 (C,COOCH₃), 54.9 (CH, C-2), 55.4 (ArOCH₃), 55.5 (ArOCH₃), 57.9 (CH, C-6), 98.7 (C, C-3), 113.4, 113.9, 114.4, 116.6, 126.0, 126.2, 127.8, 128.1, 129.3, 129.7 (ArCH), 135.0, 136.2, 138.4, 147.4, 156.6, 158.5, 159.1 (6ArC and C-4), 168.8 (CO); ESMS (*m/z*): [M+H]⁺, 519.2; Elemental analysis for C₃₃H₃₂N₂O₄: Calcd. C, 76.13; H, 6.20; N, 5.38. Found: C, 76.14; H, 6.19; N, 5.37.

4.1.2. Methyl 2,6-bis(4-bromophenyl)-1-(4-chlorophenyl)-4-(4-chlorophenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (2): White solid, mp 160-163 °C; R_f = 0.5 (Ethyl acetate/Hexane = 1/4); FTIR (KBr): ν 3741, 3618, 3020, 2360, 1652, 1614, 1497, 1216 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 2.68 (dd, *J* = 15.0 Hz and 1.95 Hz, 1H, H-5a), 2.81 (dd, *J* = 15.2 Hz and 5.6 Hz, 1H, H-5b), 3.95 (s, 3H, OMe), 5.05 (bs, 1H, H-6), 6.26-6.37 (m, 5H, 4ArH and H-2), 6.98-7.03 (m, 4H, ArH), 7.11-7.28 (m, 4H, ArH), 7.39-7.44 (m, 4H, ArH), 10.23 (bs, 1H, NH) ppm; ¹³C NMR (50 Hz, CDCl₃+CCl₄): δ 33.9 (CH₂, C-5), 51.7 (C,COOCH₃), 55.3 (CH, C-2), 57.9 (CH, C-6), 98.3 (C, C-3), 114.4 (ArCH), 121.0, 121.7 and 122.5 (3ArC), 127.3, 128.4, 128.6, 129.4, 129.6, 131.9, 132.3 (ArCH), 132.5, 136.5, 141.2, 142.4, 145.2, 155.6 (8ArC and C-4), 168.4 (CO); ESMS (*m/z*): [M+H]⁺, 663; Elemental analysis for C₃₁H₂₄Br₂Cl₂N₂O₂: Calcd. C, 67.23; H, 5.13; N, 4.75. Found: C, 67.24; H, 5.12; N, 4.77.

4.1.3. Methyl 2,6-bis(4-fluorophenyl)-1-phenyl-4-(phenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (3): White solid, mp 160 °C; R_f = 0.5 (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3738, 3681, 3621, 3020, 1652, 1652, 1595, 1504, 1372, 1323, 1216 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 2.78 (dd, *J* = 14.9 Hz and 1.56 Hz, 1H, H-5a), 2.87 (dd, *J* = 15.1 Hz and 5.3 Hz, 1H, H-5b), 3.97 (s, 3H, OMe), 5.10 (bs, 1H, H-6), 6.40-6.51 (m, 5H, 4ArH and H-2), 6.63-6.68 (t, *J* = 7.1 Hz, 1H, ArH), 6.85-7.26 (m, 13H, ArH), 10.32 (bs, 1H, NH) ppm; ¹³C NMR (50 MHz, CDCl₃+CCl₄): δ 34.2 (CH₂, C-5), 51.5 (C,COOCH₃), 55.0 (CH, C-2), 57.7 (CH, C-6), 98.2 (C, C-3), 113.4, 115.3, 115.7, 116.1, 117.1, 126.2, 126.4, 128.2, 128.4, 128.5, 128.6, 129.4, 129.5, 129.8 (ArCH), 138.1, 138.5, 139.7, 146.9, 156.4, 159.5, 164.4 (6ArC and C-4), 168.6 (CO); ESMS (*m/z*): [M+H]⁺, 496.

4.1.4. Methyl 2,6-bis(4-fluorophenyl)-1-(4-methoxyphenyl)-4-(4-methoxyphenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (4): White solid, mp 205 °C; R_f = 0.5 (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3677, 3416, 3062, 2362, 1654, 1607, 1510, 1456, 1243 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 2.62 (dd, *J* = 15.2 Hz and 2.6 Hz, 1H, H-5a), 2.73 (dd, *J* = 15.2 Hz and 5.3 Hz, 1H, H-5b), 3.68 (s, 3H, ArOMe), 3.77 (s, 3H, ArOMe), 3.96 (s, 3H, OMe), 4.97 (bs, 1H, H-6), 6.19 (s, 1CH, H-2), 6.35-6.38 (m, 4H, ArH), 6.61-6.68 (m, 4H, ArH), 6.92-7.01 (m, 4H, ArH), 7.09-7.14 (m, 2H, ArH), 7.19-7.24 (m, 2H, ArH), 10.18 (bs, 1H, NH) ppm. ¹³C NMR (50 MHz, CDCl₃+CCl₄): δ 34.1 (CH₂, C-5), 51.3 (C, COOCH₃), 55.6 (CH, C-2), 55.6 (ArOCH₃), 55.8 (ArOCH₃), 57.6 (CH, C-6),

97.1 (C, C-3), 114.5, 114.9, 115.1, 115.5, 115.6, 116.0, 128.1, 128.4, 128.6, 128.1 (ArCH), 130.9, 139.0, 140.0, 141.5, 151.8, 157.1, 158.3, 159.5, 164.4 (8ArC and C-4), 168.4 (CO); ESMS (m/z): $[M+H]^+$, 530.

4.1.5. Methyl 2,6-bis(4-bromophenyl)-1-phenyl-4-(phenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (5): White solid, mp 140 °C; $R_f = 0.5$ (Ethyl acetate/Hexane = 1/4); FTIR (KBr): ν 3685, 3395, 3020, 2361, 1653, 1595, 1500, 1427, 1216, cm^{-1} ; $^1\text{H NMR}$ (300 MHz, $\text{CDCl}_3 + \text{CCl}_4$): δ 2.77 (dd, $J = 15.0$ Hz and 2.25 Hz, 1H, H-5a), 2.86 (dd, $J = 15.0$ Hz and 5.31 Hz, 1H, H-5b), 3.96 (s, 3H, OMe), 5.06 (bs, 1H, H-6), 6.38-6.70 (m, 5H, 4ArH and H-2), 6.76-6.85 (m, 1H, ArH), 7.02-7.55 (m, 9H, ArH), 7.69-7.71 (d, $J = 6.99$ Hz, 4H, ArH), 10.29 (bs, 1H, NH) ppm; $^{13}\text{C NMR}$ (50 Hz, $\text{CDCl}_3 + \text{CCl}_4$): δ 34.0 (CH_2 , C-5), 51.6 (OCH_3), 55.2 (CH, C-2), 57.8 (CH, C-6), 97.9 (C, C-3), 113.3, 117.2 (ArCH), 120.7, 121.4 (ArC and C-4) 126.2, 126.5, 128.5, 128.9, 129.5, 131.8, 132.1 (ArCH), 138.0, 141.9, 143.3, 146.9, 156.4 (ArC), 168.7 (CO); ESMS (m/z): $[M+H]^+$, 622; Elemental analysis for $\text{C}_{31}\text{H}_{26}\text{N}_2\text{O}_2\text{Br}_2$: Calcd. C, 60.21; H, 4.24; N, 4.53. Found: C, 60.22; H, 4.23; N, 4.52.

4.1.6. Methyl 2,6-bis(3-chlorophenyl)-1-(4-methoxyphenyl)-4-(4-methoxyphenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (6): White solid, mp 160 °C; $R_f = 0.5$ (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3781, 3374, 3019, 2361, 1652, 1594, 1511, 1465, 1371, 1216 cm^{-1} ; $^1\text{H NMR}$ (300 MHz, $\text{CDCl}_3 + \text{CCl}_4$): δ 2.64 (dd, $J = 15.0$ Hz and 2.37 Hz, 1H, H-5a), 2.75 (dd, $J = 15.2$ Hz and 5.4 Hz, 1H, H-5b), 3.68 (s, 3H, ArOMe), 3.77 (s, 3H, ArOMe), 3.91 (s, 3H, OMe), 4.97 (bs, 1H, H-6), 6.19-6.38 (m, 5H, 4ArH and H-2), 6.61-6.68 (m, 4H, ArH), 6.92-7.24 (m, 8H, ArH), 10.18 (bs, 1H, NH) ppm; $^{13}\text{C NMR}$ (50 Hz, $\text{CDCl}_3 + \text{CCl}_4$): δ 33.4 (CH_2 , C-5), 50.9 (OCH_3), 55.1 (CH, C-2), 55.2 (ArOCH_3), 57.5 (CH, C-6), 114.0, 114.4, 114.7, 124.8, 124.9, 126.6, 126.9, 127.3, 128.0, 129.8, 130.0 (ArCH), 134.3, 134.5, 140.6, 145.2, 146.4, 151.6, 156.4, 158.1 (ArC and C-4), 167.9 (CO); ESMS (m/z): $[M+H]^+$, 562; Elemental analysis for $\text{C}_{33}\text{H}_{30}\text{N}_2\text{O}_4\text{Cl}_2$: Calcd. C, 67.23; H, 5.13; N, 4.75. Found: C, 67.24; H, 5.12; N, 4.76.

4.1.7. Methyl 1,2,6-tris(4-methoxyphenyl)-4-(4-methoxyphenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (7): White solid, mp 158-160 °C; $R_f = 0.5$ (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3678, 3437, 3020, 2360, 1649, 1512, 1461, 1217 cm^{-1} ; $^1\text{H NMR}$ (300 MHz, $\text{CDCl}_3 + \text{CCl}_4$): δ 2.64 (dd, $J = 15.0$ Hz and 2.4 Hz, 1H, H-5a), 2.77 (dd, $J = 15.0$ Hz and 5.4 Hz, 1H, H-5b), 3.68 (s, 3H, ArOMe), 3.77 (s, 3H, ArOMe), 3.79 (s, 3H, ArOMe), 3.82 (s, 3H, ArOMe), 3.91 (s, 3H, OMe), 4.98 (bs, 1H, H-6), 6.22 (s, 1CH, H-2), 6.30 (d, $J = 7.6$ Hz, 2H, ArH), 6.43 (d, $J = 9.0$ Hz, 2H, ArH), 6.64 (d, $J = 7.8$ Hz, 4H, ArH), 6.79-6.84 (m, 4H, ArH), 7.06 (d, $J = 8.5$ Hz, 2H, ArH), 7.17 (d, $J = 8.5$ Hz, 2H, ArH), 10.16 (bs, 1H, NH) ppm; $^{13}\text{C NMR}$ (50 MHz, $\text{CDCl}_3 + \text{CCl}_4$): δ 34.1 (CH_2 , C-5), 51.2 (OCH_3), 55.4 (CH, C-2), 55.5 (ArOCH_3), 55.6 (ArOCH_3), 55.9 (ArOCH_3), 57.8 (CH, C-6), 97.5 (C, C-3), 113.8, 113.9, 114.3, 114.3, 114.8, 128.0, 128.2, 128.5 (ArCH), 131.2, 135.5, 136.5, 141.9, 151.4, 157.4, 158.1, 158.4, 159.0 (8ArC and C-4), 168.9 (CO); ESMS (m/z): $[M+H]^+$, 580; Elemental Analysis for $\text{C}_{35}\text{H}_{36}\text{N}_2\text{O}_6$, 72.39; H, 6.25; N, 4.82. Found: C, 72.36; H, 6.26; N, 4.81.

4.1.8. Methyl-(4-chlorophenyl)-4-(4-chlorophenylamino)-2,6-bis(methoxyphenyl)-1,2,5,6-tetrahydropyridine-3-carboxylate (8): White solid, mp 195 °C; $R_f = 0.5$ (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3400, 3020, 2360, 1652, 1603, 1500, 1432, 1216; $^1\text{H NMR}$ (300 MHz, CDCl_3): δ 2.71 (dd, $J = 15.0$ Hz and 5.5 Hz, 1H, H-5a), 2.85 (dd, $J = 15.0$ Hz and 2.1 Hz, 1H, H-5b), 3.80 (s, 3H, ArOMe), 3.82 (s, 3H, ArOMe), 3.95 (s, 3H, OMe), 5.06 (bs, 1H, H-6), 6.26-6.32 (m, 3H, ArH and H-2), 6.44-6.47 (d, $J = 9.0$ Hz, 2H, ArH), 6.82-6.86 (m, 4H, ArH), 7.00-7.20 (m, 8H, ArH), 10.24 (bs, 1H, NH) ppm; $^{13}\text{C NMR}$ (50 Hz, CDCl_3): δ 34.0 (CH_2 , C-5), 51.6 (OCH_3), 55.1 (CH, C-2), 55.2 (ArOCH_3), 55.5 (ArOCH_3), 57.9 (CH, C-6), 97.9 (C, C-3), 113.3, 117.2 (ArCH), 120.7, 121.4 (ArC), 126.2, 126.5, 128.5, 128.8, 129.5 (ArCH), 131.8, 132.1, 138.0, 141.8, 143.3, 146.8, 156.4 (ArC and C-4), 168.7 (CO); ESMS (m/z): $[M+H]^+$, 588; Elemental analysis for $\text{C}_{33}\text{H}_{30}\text{N}_2\text{O}_4\text{Cl}_2$: Calcd. C, 67.23; H, 5.13; N, 4.75. Found: C, 67.25; H, 5.11; N, 4.78.

4.1.9. Methyl-1-(4-bromophenyl)-4-(4-bromophenylamino)-2,6-bis(4-methoxyphenyl)-1,2,5,6-tetrahydropyridine-3-carboxylate (9): White solid, mp 178 °C; $R_f = 0.5$ (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3781, 3374, 3019, 2361, 1652, 1594, 1511, 1465, 1371, 1216 cm^{-1} ; $^1\text{H NMR}$ (300 MHz, $\text{CDCl}_3 + \text{CCl}_4$): δ 2.69 (d, $J = 14.9$ Hz, 1H, H-5a), 2.83 (dd, $J = 14.9$ Hz and 5.4 Hz, 1H, H-5b), 3.78 (s, 3H, ArOMe), 3.81 (s, 3H, ArOMe), 3.96 (s, 3H, OMe), 5.04 (bs, 1H, H-6), 6.26-6.29 (m, 3H, ArH and H-2), 6.42 (d, $J = 8.1$ Hz, 2H, ArH), 6.80-6.85 (m, 4H, ArH), 6.98-7.17 (m, 8H, ArH), 10.28 (bs, 1H, NH) ppm; $^{13}\text{C NMR}$ (50 Hz, $\text{CDCl}_3 + \text{CCl}_4$): δ 34.0 (CH_2 , C-5), 51.6 (OCH_3), 55.1 (CH, C-2), 57.8 (CH, C-6), 61.2 (2ArOCH_3), 114.4, 115.4, 115.8, 116.3, 116.5 (ArCH), 122.4 (ArC), 127.2, 128.1, 128.3, 128.4, 128.8, 129.3, 129.5, 129.9 (ArCH), 132.1, 136.6, 137.8, 138.9, 145.3, 155.5, 159.6 (ArC), 164.5 (ArC), 168.2 (CO); ESMS (m/z): $[M+H]^+$, 562; Elemental analysis for $\text{C}_{33}\text{H}_{30}\text{N}_2\text{O}_4\text{Br}_2$: Calcd. C, 58.42; H, 4.46; N, 4.13. Found: C, 58.51; H, 4.48; N, 4.11.

4.1.10. Methyl-(4-chlorophenyl)-4-(4-chlorophenylamino)-2,6-bis(4-fluorophenyl)-1,2,5,6-tetrahydropyridine-3-carboxylate (10): White solid, mp 176 °C; $R_f = 0.5$ (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3697, 3627, 3021, 2360, 1655, 1600, 1499, 1216 cm^{-1} ; $^1\text{H NMR}$ (300 MHz, $\text{CDCl}_3 + \text{CCl}_4$): δ 2.70 (dd, $J = 14.8$ Hz and 1.95 Hz, 1H, H-5a), 2.82 (dd, $J = 15.0$ Hz and 5.3 Hz, 1H, H-5b), 3.96 (s, 3H, OMe), 5.07 (bs, 1H, H-6), 5.94-6.35 (m, 5H, 4ArH and H-2), 6.90-7.34 (m, 12H, ArH), 10.29 (bs, 1H, NH) ppm; $^{13}\text{C NMR}$ (50 Hz, $\text{CDCl}_3 + \text{CCl}_4$): δ 34.0 (CH_2 , C-5), 51.3

(OCH₃), 55.5 (CH, C-2), 57.6 (CH, C-6), 97.1 (C, C-3), 114.4, 114.9, 115.1, 115.5, 115.6, 116.0, 128.1, 128.4, 128.5, 128.6, 128.8 (ArCH), 130.9, 139.0, 139.8, 141.3, 151.9, 156.8, 158.3, 159.4, 164.3 (8ArC and C-4), 168.4 (CO); ESMS (*m/z*): [M+H]⁺, 565.

4.1.11. Methyl 2,6-bis(4-(benzyloxy)phenyl)-1-phenyl-4-(phenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (11): White solid, mp 155 °C; *R_f* = 0.5 (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3654, 3450, 3032, 2363, 1656, 1592, 1504, 1453, 1378, 1246 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 2.74-2.79 (m, 1H, H-5a), 2.84-2.91 (dd, *J* = 15.0 Hz and 5.2 Hz, 1H, H-5b), 3.97 (s, 3H, OMe), 4.99- 5.18 (m, 5H, 2 OCH₂, H-6), 6.38-6.56 (m, 5H, 4ArH and H-2), 6.63 (t, *J* = 7.1Hz, 1H, ArH), 6.90-6.93 (m, 4H, ArH), 7.09-7.11(m, 7H, ArH), 7.17-7.28 (m, 2H, ArH), 7.35-7.45 (m, 10H, ArH), 10.31 (bs, 1H, NH) ppm; ¹³C NMR (50 Hz, CDCl₃+CCl₄): δ 34.1 (CH₂, C-5), 51.3 (OCH₃), 55.0 (CH, C-2), 57.9 (CH, C-6), 70.3 and 70.4 (OCH₂), 98.5 (C, C-3), 113.4, 114.9, 115.4, 116.5, 126.0, 126.2, 127.8, 127.9, 128.1, 128.3, 128.9, 129.2, 129.7 (ArCH), 135.3, 136.4, 137.5, 137.6, 138.3, 147.3, 156.6, 157.8, 158.3 (ArC and C-4), 168.8 (CO); ESMS (*m/z*): [M+H]⁺, 672; Elemental analysis for C₄₅H₄₀N₂O₄: Calcd. C, 67.23; H, 5.13; N, 4.75. Found: C, 67.24; H, 5.15; N, 4.73.

4.1.12. Methyl 1,2,6-triphenyl-4-(phenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (12): White solid, mp 194 °C; *R_f* = 0.5 (Ethyl acetate/Hexane = 1/4); FTIR (KBr): ν 3859, 3427, 3020, 2360, 1653, 1592, 1498, 1257 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 2.73 (dd, *J* = 15.0 Hz and 2.2 Hz, 1H, H-5a), 2.85 (dd, *J* = 15.0 Hz and 5.4 Hz, 1H, H-5b), 3.91 (s, 3H, OMe), 5.1 (bs, 1H, H-6), 6.25-6.28 (m, 2H, ArH and H-2), 6.42-6.50 (m, 2H, ArH), 6.54-6.59 (t, *J* = 7.2 Hz, 1H, ArH), 6.99-7.09 (m, 5H, ArH), 7.13-7.29 (m, 10H, ArH), 10.29 (bs, 1H, NH) ppm; ¹³C NMR (50 Hz, CDCl₃+CCl₄): δ 34.0 (CH₂, C-5), 51.4 (OCH₃), 55.5 (CH, C-2), 58.7 (CH, C-6), 98.5 (C, C-3), 113.4, 116.7, 126.1, 126.2, 126.8, 127.0, 127.6, 128.7, 129.0, 129.2, 129.3, 129.7 (ArCH), 138.3, 143.2, 144.3, 147.3, 156.5 (4ArC and C-4), 168.9 (CO); ESMS (*m/z*): [M+H]⁺, 460; Elemental Analysis for C₃₁H₂₈N₂O₂: C, 80.84; H, 6.13; N, 6.08. Found: C, 80.82; H, 6.15; N, 6.06.

4.1.13. Methyl 2,6-bis(3-chlorophenyl)-1-phenyl-4-(phenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (13): White solid, mp 220 °C; *R_f* = 0.5 (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3680, 3398, 3020, 2360, 1652, 1593, 1502, 1427, 1261, 1216 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 2.82 (dd, *J* = 15.3 Hz and 2.1 Hz, 1H, H-5a), 2.85 (dd, *J* = 15.2 Hz and 5.3 Hz, 1H, H-5b), 3.97 (s, 3H, OMe), 5.13 (bs, 1H, H-6), 6.39-6.49 (m, 5H, 4ArH and H-2), 6.65-6.69 (t, *J* = 7.2 Hz, 1H, ArH), 7.03-7.31 (m, 14H, ArH), 10.29 (bs, 1H, NH) ¹³C NMR (50 Hz, CDCl₃+CCl₄): δ 34.1 (CH₂, C-5), 51.5 (OCH₃), 55.3 (CH, C-2), 57.9 (CH, C-6), 97.6 (C, C-3), 113.3, 117.4, 125.0, 126.5, 126.6, 126.8, 127.1, 127.9, 129.4, 129.5, 129.8, 130.3(ArCH), 134.8, 135.0, 137.9, 145.1, 146.5, 146.7, 156.3 (6ArC and C-4), 168.4 (CO); ESMS (*m/z*): [M+H]⁺, 530; Elemental analysis for C₃₃H₂₆N₂O₂Cl₂: Calcd. C, 70.32; H, 4.95; N, 4.61. Found: C, 70.34; H, 4.93; N, 4.62.

4.1.14. Ethyl 2,6-bis(3-chlorophenyl)-1-(4-methoxyphenyl)-4-(4-methoxyphenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (14): Pale yellow solid, mp 167-170 °C, *R_f* = 0.5 (Ethyl acetate/Hexane = 1/4); FTIR (KBr): ν 3447, 2987, 2836, 2362, 1646 cm⁻¹; ¹H NMR (300 MHz, CDCl₃:CCl₄): δ 1.47 (t, *J* = 7.1 Hz, 3H, COOCH₂CH₃), 2.64 (dd, *J* = 15.1 Hz and 2.7 Hz, 1H, H-5a), 2.77 (dd, *J* = 15.1 Hz and 5.4Hz, 1H, H-5b), 3.79 and 3.68 (two s, 6H, 2xOMe), 4.51-4.29 (dq, *J* = 7.1 Hz and 3.7Hz, 2H, OCH₂), 5.01 (bs, 1H, H-6), 6.24-6.41 (m, 5H, 4ArH and H-2), 6.67-6.71 (m, 4H, ArH), 7.03-7.35 (m, 8H, ArH), 10.17 (bs, 1H, NH); ¹³C NMR (50 MHz, CDCl₃+CCl₄): δ 15.2 (CH₃, COOCH₂CH₃), 33.9 (CH₂, C-5), 55.7 and 55.9 (OCH₃), 56.1(CH, C-2), 57.9 (CH, C-6), 60.0 (OCH₂), 96.9 (C, C-3), 114.4, 114.8, 115.0 (ArCH), 125.1, 125.2, 127.0, 127.4, 127.7, 128.4, 129.7 (ArCH), 130.8, 134.7, 134.9, 141.2, 145.6, 147.0, 151.9, 156.8, 158.4 (8ArC and C-4), 168.2 (CO); ESMS (*m/z*): [M+H]⁺, 602; Elemental Analysis for C₃₄H₃₂Cl₂N₂O₄: C, 67.66; H, 5.34; N, 4.64; Found: C, 67.64; H, 5.36; N, 4.63.

4.1.15. Ethyl-(4-chlorophenyl)-4-(4-chlorophenylamino)-2,6-di-thiophen-2-yl)-1,2,5,6-tetrahydropyridine-3-carboxylate (15): White solid, mp 217 °C, *R_f* = 0.5 (Ethyl acetate/Hexane = 1/4); FTIR (KBr): ν 3780, 3019, 2926, 2361, 1651 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 1.42-1.49 (t, *J* = 7.1 Hz, 3H, COOCH₂CH₃), 2.83 (dd, *J* = 15.2 Hz and 2.5 Hz, 1H, H-5a), 3.1 (dd, *J* = 15.2 Hz and 5.1 Hz, 1H, H-5b), 4.25-4.49 (m, 2H, OCH₂), 5.3 (bs, 1H, H-6), 6.36 (s, 1H, H-2), 6.44 (d, *J* = 12.9 Hz, 2H, ArH), 6.63 (d, *J* = 13.6 Hz, 2H, ArH), 6.81-7.02 (m, 4H, ArH), 7.0-7.26 (m, 6H, ArH), 10.40 (bs, 1H, NH) ppm; ¹³C NMR (50 MHz, CDCl₃): δ 15.3 (CH₃, COOCH₃), 34.5 (CH₂, C-5), 53.0 (CH, C-2), 54.2 (CH, C-6), 60.2 (OCH₂), 98.4 (C, C-3), 109.9, 114.8(ArCH), 122.6 (ArC), 124.1, 124.5, 124.8, 126.9, 127.2 (ArCH), 129.1, 129.6 (ArCH), 131.7, 136.9, 145.0, 147.1, 148.7, 155.7 (ArC and C-4), 168.0 (CO); ESMS (*m/z*): [M+H]⁺, 554; Elemental Analysis for C₂₈H₂₄Cl₂N₂O₂S₂: C, 60.54; H, 4.35; N, 5.04. Found: C, 60.56; H, 4.34; N, 5.03.

4.1.16. Methyl 1-(4-methoxyphenyl)-4-(4-methoxyphenylamino)-2,6-di-thiophen-2-yl)-1,2,5,6-tetrahydropyridine-3-carboxylate (16): White solid, mp 210 °C; *R_f* = 0.5 (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3694, 3020, 2361, 1655, 1596, 1497, 1437, 1216 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): 2.8-3.0 (m, 2H, H-5a and H-5b), 3.71 (s, 3H, ArOMe), 3.79 (s, 3H, ArOMe), 3.89 (s, 3H, OMe), 5.27 (bs, 1H, H-6), 5.95 (s, 1H, H-2), 6.25-7.27 (m, 14 H, ArH), 10.48 (bs, 1H, NH) ppm; ¹³C NMR (50 Hz, CDCl₃+CCl₄): δ 33.3 (CH₂, C-5), 51.1 (OCH₃), 54.2

(CH, C-2), 55.6 (ArOCH₃), 55.7 (ArOCH₃), 57.3 (CH, C-6), 97.4 (C, C-3), 114.5, 114.7, 114.9, 116.6, 119.6, 122.6, 123.8, 124.1, 124.5, 124.7 (ArCH), 125.3 (ArC), 126.4, 126.7, 126.8, 127.3, 128.0 (ArCH), 131.3, 140.9, 148.0, 149.9, 154.1, 156.9 (5ArC and C-4), 168.2 (CO), ESMS (*m/z*): [M+H]⁺, 531.2; Elemental analysis for C₂₉H₂₈N₂O₄S₂: C, 65.39; H, 5.30; N, 5.26. Found: C, 65.37; H, 5.32; N, 5.24.

4.1.17. Ethyl 1,2,6-tris-(4-chlorophenyl)-4-(4-chlorophenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (17): White solid, mp 190 °C, *R_f* = 0.5 (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3780, 3019, 2926, 2361, 1651 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CD₃OD): δ 1.47-1.54 (t, *J* = 7.0 Hz, 3H, COOCH₂CH₃), 2.76 (dd, *J* = 15.1 Hz and 2.2 Hz, 1H, H-5a), 2.82 (dd, *J* = 15.2 Hz and 5.2 Hz, 1H, H-5b), 4.26-4.54 (m, 2H, OCH₂), 5.06 (bs, 1H, H-6), 6.25-6.42 (m, 5H, 4ArH and H-2), 6.96-7.30 (m, 12H, ArH), 10.25 (bs, 1H, NH); ¹³C NMR (50 MHz, CDCl₃+CD₃OD): δ 19.1 (CH₃, COOCH₂CH₃), 37.6 (CH₂, C-5), 59.4 (CH, C-2), 61.8 (CH, C-6), 64.3 (OCH₂), 102.3 (C, C-3), 118.3 (ArCH), 126.5 (ArC), 128.6, 128.8 (ArCH), 130.2 (C-4), 130.7, 131.0, 131.2, 131.3, 132.0, 132.9, 133.2, 133.4, 133.8, 134.3 (ArCH), 136.2, 138.9, 139.1, 140.3, 148.5, 149.0, 149.8 (ArC), 172.0 (CO); ESMS (*m/z*): [M+H]⁺, 611; Elemental Analysis for C₃₂H₂₆Cl₄N₂O₂: C, 62.76; H, 4.16; N, 4.57. Found: C, 62.78; H, 4.14; N, 4.56.

4.1.18. Ethyl-1-(4-methoxyphenyl)-4-(4-methoxyphenylamino)-2,6-diphenyl-1,2,5,6-tetrahydropyridine-3-carboxylate (18): White solid, mp 173 °C, *R_f* = 0.5 (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3751, 3447, 3230, 2991, 2362, 1646, 1600, 1511 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 1.41-1.45 (t, *J* = 7.0 Hz, 3H, COOCH₂CH₃), 2.62 (dd, *J* = 15.2 Hz and 2.2 Hz, 1H, H-5a), 2.76 (dd, *J* = 15.1 Hz and 5.6 Hz, 1H, H-5b), 3.63 (s, 3H, ArOMe), 3.71 (s, 3H, ArOMe), 4.23-4.48 (m, 2H, OCH₂), 5.02 (bs, 1H, H-6), 6.18-6.21 (d, *J* = 8.64 Hz, 2H, ArH), 6.29 (s, 1H, H-2), 6.39-6.42 (d, *J* = 9.0 Hz, 2H, ArH), 6.56-6.63 (m, 4H, ArH), 7.14-7.29 (m, 10H, ArH), 10.30 (bs, 1H, NH) ppm; ¹³C NMR (50 MHz, CDCl₃+CCl₄): δ 15.2 (CH₃, COOCH₂CH₃), 34.0 (CH₂, C-5), 55.6 (CH, C-2), 55.9 (ArOCH₃), 56.2 (ArOCH₃), 58.7 (CH, C-6), 59.8 (OCH₂), 97.7 (C, C-3), 114.3, 114.6, 114.9, 126.6, 127.0, 127.2, 127.4, 128.2, 128.5, 129.0 (ArCH), 131.2, 141.9, 143.7, 144.7, 151.5, 157.0, 158.2 (6ArC and C-4), 168.2 (CO); ESMS (*m/z*): [M+H]⁺, 611; Elemental Analysis for C₃₄H₃₄N₂O₄: C, 76.38; H, 6.41; N, 5.24. Found: C, 76.36; H, 6.43; N, 5.22.

4.1.19. Ethyl 2,6-bis(phenyl)-1-phenyl-4-(phenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (19): Pale yellow solid, mp 202 °C, *R_f* = 0.5 (Ethyl acetate/Hexane = 1/4); FTIR (KBr): ν 3446, 3244, 2974, 2361, 1646 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 1.52 (t, *J* = 7.1 Hz, 3H, COOCH₂CH₃), 2.73 (dd, *J* = 15.0 Hz and 2.0 Hz, 1H, H-5a), 2.89 (dd, *J* = 15.0 Hz and 5.7 Hz, 1H, H-5b), 4.54-4.34 (dq, *J* = 7.1 Hz and 3.7 Hz, 2H, OCH₂), 5.13 (bs, 1H, H-6), 6.20-6.43 (m, 3H, ArH and H-2), 7.00-7.09 (m, 4H, ArH), 7.17-7.35 (m, 10H, ArH), 10.29 (bs, 1H, NH); ¹³C NMR (50 MHz, CDCl₃): δ 15.2 (CH₃, COOCH₃), 33.8 (CH₂, C-5), 55.7 (CH, C-2), 58.7 (CH, C-6), 60.2 (OCH₂), 99.2 (C, C-3), 114.4 (ArCH), 121.8 (ArC), 126.7, 126.9, 127.0, 127.3, 127.8, 128.7, 129.1, 129.2, 129.4 (ArCH), 131.8, 136.8, 142.7, 143.6, 145.8, 155.6 (5ArC and C-4), 168.3 (CO); ESMS (*m/z*): [M+H]⁺, 543; Elemental Analysis for C₃₂H₂₈Cl₂N₂O₂: C, 70.72; H, 5.19; N, 5.15. Found: C, 70.71; H, 5.17; N, 5.18.

4.1.20. Ethyl-1-phenyl-4-(phenylamino)-2,6-di(pyridin-3-yl)-1,2,5,6-tetrahydropyridine-3-carboxylate (20): White solid, mp 178 °C, *R_f* = 0.5 (methanol/chloroform = 1/2); FTIR (KBr): ν 3368, 3244, 2972, 2362, 1720, 1654 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 1.46-1.50 (t, *J* = 7.0 Hz, 3H, COOCH₂CH₃), 2.83-2.84 (d, *J* = 3.8 Hz, 2H, H-5), 4.30-4.50 (m, 2H, OCH₂), 5.20 (bs, 1H, H-6), 6.42-6.48 (m, 5H, 4ArH and H-2), 6.66 (t, *J* = 7.2 Hz, 1H, ArH), 7.06-7.28 (m, 7H, ArH), 7.43-7.45 (d, *J* = 7.7 Hz, 1H, ArH), 7.58-7.61 (d, *J* = 7.8 Hz, 1H, ArH), 8.43-8.53 (m, 3H, ArH), 8.64 (s, 1H, ArH), 10.38 (bs, 1H, NH); ¹³C NMR (50 MHz, CDCl₃+CCl₄): δ 15.1 (CH₃, COOCH₃), 34.1 (CH₂, C-5), 53.9 (CH, C-2), 55.9 (C-6 and OCH₃), 60.4 (OCH₂), 97.6 (C, C-3), 113.5, 117.7, 123.4, 123.7, 126.0, 126.6, 129.0, 129.6, 134.5, 134.6 (ArCH), 137.8, 138.0, 139.6, 146.4 (5ArC and C-4), 148.2, 148.7, 148.9, 149.0 (ArCH), 155.7 (ArC), 168.0 (CO); ESMS (*m/z*): [M+H]⁺, 477; Elemental Analysis for C₃₀H₂₈N₄O₂: C, 75.61; H, 5.92; N, 11.76. Found: C, 75.63; H, 5.91; N, 11.74.

4.1.21. Methyl 2,6-bis(4-bromophenyl)-1-(4-methoxyphenyl)-4-(4-methoxyphenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (21): White solid, mp 179 °C, *R_f* = 0.5 (Ethyl acetate/Hexane = 1/5); FTIR (KBr): ν 3797, 3421, 3018, 2358, 1650, 1508, 1371, 1216 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): 2.61 (dd, *J* = 15.0 Hz and 2.4 Hz, 1H, H-5a), 2.72 (dd, *J* = 15.2 Hz and 5.4 Hz, 1H, H-5b), 3.68 (s, 3H, ArOMe), 3.78 (s, 3H, ArOMe), 3.90 (s, 3H, OMe), 4.95 (bs, 1H, H-6), 6.18 (s, 1CH, H-2), 6.37 (d, *J* = 8.4 Hz, 2H, ArH), 6.64-6.70 (m, 5H, ArH), 7.00 (d, *J* = 8.1 Hz, 2H, ArH), 7.13 (d, *J* = 8.1 Hz, 2H, ArH), 7.37-7.42 (m, 5H, ArH), 10.13 (s, 1H, NH) ppm; ¹³C NMR (50 Hz, CDCl₃+CCl₄): δ 34.0 (CH₂, C-5), 51.4 (OCH₃), 55.6 (ArOCH₃), 55.8 (ArOCH₃), 56.0 (CH, C-2), 57.8 (CH, C-6), 96.8 (C, C-3), 114.5, 114.7, 115.0 (ArCH), 120.6, 121.3 (2ArC), 128.1, 128.7, 129.0 (ArCH), 130.8 (C-4), 131.6, 132.0 (ArCH), 141.2, 142.3, 143.4, 151.9, 156.9, 158.4, 168.5 (CO); ESMS (*m/z*): [M+H]⁺, 562; Elemental analysis for C₃₃H₃₀Br₂N₂O₄: Calcd. C, 58.42; H, 4.46; N, 4.13. Found: C, 58.44; H, 4.48; N, 4.11.

4.1.22. Methyl 2,6-bis(3-nitrophenyl)-1-phenyl-4-(phenylamino)-1,2,5,6-tetrahydropyridine-3-carboxylate (22): Pale yellow solid, mp 180 °C, *R_f* = 0.4 (Ethyl acetate/Hexane = 1/4); FTIR (KBr): ν 3751, 3019, 2954, 2364, 1736, 1669 cm⁻¹; ¹H NMR (300 MHz, CDCl₃+CCl₄): δ 2.90 (d, *J* = 3.69 Hz, 2H, H-5), 4.01 (s, 3H, OCH₃), 5.35 (bs, 1H, H-6), 6.41-6.50 (m, 5H, ArH and H-2), 6.70 (t, *J* = 7.26 Hz, 1H, ArH), 7.08-7.17 (m, 5H, ArH), 7.44-7.51 (m, 3H, ArH), 7.68 (d, *J* = 7.62 Hz, 1H, ArH), 7.96 (s, 1H, ArH), 8.10 (m, 2H, ArH), 8.23 (s, 1H, ArH), 10.32 (bs, 1H, NH); ¹³C NMR (50

MHz, CDCl_3): δ 30.0 (CH_3 , COOCH_3), 34.1 (CH_2 , C-5), 51.8 (CH , C-2), 55.6 (CH , C-6), 57.5 (C, C-3), 113.5, 118.3, 121.8, 121.9, 122.2, 122.8, 126.1, 127.0, 129.5, 129.6, 129.8, 129.9, 132.8 (ArCH), 155.8, 149.1, 146.7, 146.1, 144.8, 137.6 (ArC and C-4), 168.2 (CO); ESMS (m/z): $[\text{M}+\text{H}]^+$, 552; Elemental Analysis for $\text{C}_{31}\text{H}_{27}\text{N}_4\text{O}_6$: C, 67.50; H, 4.93; N, 10.16. Found: C, 67.45; H, 5.00; N, 10.18.

4.1.23. (E)-methyl 2-(3-methoxybenzylidene)-3-oxobutanoate (23): Pale yellow viscous mass, $R_f = 0.7$ (Ethyl acetate/Hexane = 1/4); FTIR (KBr): ν 3697, 3429, 3020, 2360, 1725, 1601 cm^{-1} ; ^1H NMR (300 MHz, $\text{CDCl}_3+\text{CCl}_4$): δ 2.34 (d, $J = 8.40$, CH_3), 3.86 (s, 6H, $3\times\text{OCH}_3$), 6.87-6.92 (m, 2H, ArH), 7.35-7.41 (m, 2H, ArH), 7.51 (d, $J = 29.4$ Hz, 1H, CH); ESMS (m/z): $[\text{M}+\text{H}]^+$, 235.

4.1.24. (E)-N-(3-methoxybenzylidene)aniline (24): Pale yellow solid, mp 65-67 $^{\circ}\text{C}$; $R_f = 0.8$ (Ethyl acetate/Hexane = 1/4); FTIR (KBr): ν 3605, 3020, 2360, 1597 cm^{-1} ; ^1H NMR (300 MHz, $\text{CDCl}_3+\text{CCl}_4$): δ 8.38 (s, 1H, CH), 7.87 (d, $J = 8.67$ Hz, 2H, ArH), 7.41-7.36 (m, 2H, ArH), 7.24-7.18 (m, 2H, ArH), 7.00 (d, $J = 8.67$, 2H, ArH), 3.89 (s, 3H, CH_3); ESMS (m/z): $[\text{M}+\text{H}]^+$, 212.

4.2 Biology

4.2.1 Determination of antimalarial activity in vitro¹⁶⁻¹⁸

The antimalarial activity of the compounds and reference drug (chloroquine diphosphate MIC 0.025 $\mu\text{g}/\text{ml}$) were assayed against 3D7 strain of *P. falciparum* by the schizont maturation test. Briefly drug dilutions (100 $\mu\text{L}/\text{well}$) were prepared in complete RPMI 1640 and 10 μL of parasite preparation was added to each well. The final culture suspension had a hematocrit of 3–4% and a 1.0–2.0% infection of parasitized erythrocytes (>95% rings) in culture medium containing 0.5% AlbuMax II and 15 μM hypoxanthine. Micro culture plates were incubated for 30-39 h at 37 $^{\circ}\text{C}$ in an Incubator supplied with 5% CO_2 to allow the development of malaria parasites. After incubation, culture plates were taken out and maximum supernatant medium was removed and thin blood smears of each well content were made and stained with 5% Giemsa stain. These smears were checked for the maturation of schizont relative to the controls. These tests, measure the drug sensitivity of *P. falciparum* following the WHO standard protocol with minor modifications for the assessment of the inhibition of schizont maturation.

Acknowledgements

Authors thank ICMR and DBT New Delhi for financial assistance. VP and JP thank CSIR New Delhi for JRF and SRF respectively. It is a CDRI communication No.7466.

References and notes

- Tripathi, R. P.; Mishra, R. C.; Dwivedi, N.; Tewari, N.; Verma, S. S. *Curr. Med. Chem.* **2005**,
- Breman, J.; Egan, A.; Keusch, G. *Am. J. Trop. Med. Hyg.* **2001**, *64*(Suppl.1–2), iv–vii.
- Snow, R. W.; Craig, M.; Deichmann, U.; Marsh, K. *Bull. W. H. O.* **1999**, *77*, 624–640.
- White, N. J. Antimalarial drug resistance. *J. Clin. Invest.* **2004**, *113*, 1084–1092.
- (a) Desideri, N.; Galli, A.; Sestili, J.; Stein, M. L. *Arch. Pharm.*, **1992**, *325*, 29; (b) Kobayashi, S.; Ueno, M.; Suzuki, R.; Ishitani, H.; Kim, H. S.; Wataya, Y. *J. Org. Chem.*, **1999**, *64*, 6833; (c) Harrison, T.; Williams, B. J.; Swain, C. J.; Ball, R. G. *Bioorg. Med. Chem. Lett.*, **1994**, *4*, 2545; (d) Desai, M. C.; Lefkowitz, S. L.; Thadeio, P. F.; Longo, K. P.; Snider, R. M. *J. Med. Chem.*, **1992**, *35*, 4911; (e) Watson, P. S.; Jiang, B.; Scott, B. *Org. Lett.*, **2000**, *2*, 3679; (f) Elbein, A. D.; Molyneux, R. In: Pelletier, S. W. Editor, *Alkaloids; Chemical and Biological Perspectives*, John Wiley and Sons, New York **1987**, *57*, p. 1; (g) Pinder, A. R. *Nat. Prod. Rep.*, **1992**, *9*, 491.
- (a) Beeler, A. B.; Gadepalli, R. S. V. S.; Steyn, S.; Castagnoli Jr, N.; Rimoldi, J. M. *Bioorg. Med. Chem.*, **2003**, *11*, 5229; (b) Deskus, J. A.; Epperson, J. R.; Charles, P. S.; Joseph, A. C.; Dextraze, P.; Qian-Cutrone, J.; Gao, Q.; Ma, B.; Beno, B. R.; Mattson, G. K.; Molski, T. F.; Krause, R. G.; Taber, M. T.; Lodge, N. J.; Mattson, R. J. *Bioorg. Med. Chem. Lett.*, **2007**, *17*, 3099; (c) Seyfried, C. A.; Greiner, H. E.; Haase, A. F. *Eur. J. Pharmacol.*, **1989**, *160*, 31; (d) Jaen, J. C.; Wise, L. D.; Heffner, T. G.; Pugsley, T. A.; Meltzer, L. T. *J. Med. Chem.* **1988**, *31*, 1621; (e) Jaen, J. C.; Caprathe, B. W.; Wise, L. D.; Meltzer, L. T.; Pugsley, T. A.; Heffner, T. G. *Bioorg. Med. Chem. Lett.*, **1993**, *3*, 639.
- Gwaltney II, S. L.; O'Connor, S. J.; Nelson, L. T. J.; Sullivan, G. M.; Imade, H.; Wang, W.; Hasvold, L.; Li, Q.; Cohen, J.; Gu, W. -Z.; Tahir, S. K.; Bauch, J.; Marsh, K.; Ng, S. -C.; Frost, D. J.; Zhang, H.; Muchmore, S.; Jakob, C. G.; Stoll, V.; Hutchins, C. *Bioorg. Med. Chem. Lett.*, **2003**, *13*, 1359.
- (a) Kamei, K.; Maeda, N.; Katsuragi-Ogino, R.; Koyama, M.; Nakajima, M.; Tatsuoka, T.; Ohno, T.; Inoue, T. *Bioorg. Med. Chem. Lett.*, **2005**, *15*, 2990; (b) Van Voorhis, W. C.; Rivas, K. L.; Bendale, P.; Nallan, L.; Hornéy, C.; Barrett, L. K.; Bauer, K. D.; Smart, B. P.; Ankala, S.; Hucke, O.; Verlinde, C. L. M. J.; Chakrabarti, D.; Strickland, C.;

- Yokoyama, K.; Buckner, F. S.; Hamilton, A. D.; Williams, D. K.; Lombardo, L. J.; Floyd, D.; Gelb, M. H. *J. Med. Chem.*, **2005**, *48*, 3704.
9. For a review on asymmetric synthesis of piperidines, see: (a) Bailey, P. D.; Millwood, P. A.; Smith, P. D. *Chem. Commun.*, **1998**, 633; (b) Laschat, S.; Dickner, T. *Synthesis*, **2000**, 1781; (c) Buffat, M. G. P. *Tetrahedron*, **2004**, *60*, 1701; (d) Weintraub, P. M.; Sabol, J. S.; Kane, J. M.; Borcharding, D. R. *Tetrahedron*, **2003**, *59*, 2953; (e) Felpin, F. X.; Lebreton, J. *Eur. J. Org. Chem.*, **2003**, 3693; (f) Barluenga, J.; Mateos, C.; Aznar, F.; Valdes, C. *J. Org. Chem.*, **2004**, *69*, 7114; (g) Barluenga, J.; Mateos, C.; Aznar, F.; Valdes, C. *Org. Lett.*, **2002**, *4*, 1971; (h) Badorrey, R.; Cattivola, C.; Diaz-De-Villegas, M. D.; Galvez, J. A. *Tetrahedron*, **1999**, *55*, 7601; (i) Barluenga, J.; Aznar, F.; Ribas, C.; Valdes, C. *J. Org. Chem.*, **1999**, *64*, 3736; (j) Heintzelman, G. R.; Weinreb, S. M.; Parvez, M. *J. Org. Chem.*, **1996**, *61*, 4594; (k) Waldmann, H. *Synthesis*, **1994**, 535.
10. For a recent review on dihydropyridines, see: (a) Lavilla, R. *J. Chem. Soc., Perkin Trans. 1*, **2002**, 1141; (b) Viso, A.; Pradilla, R. F. D. I.; Garcia, A.; Flores, A. *Chem. Rev.*, **2005**, *105*, 3167; (c) Davis, F. A.; Zhang, Y.; Li, D. *Tetrahedron Letters*, **2007**, *48*, 7838; (d) Xiao, D.; Wang, L.; Feng, X. *Synlett*, **2005**, *10*, 1531; (e) Baliah, V.; Jeyraman, R.; Chandrasekaran, L. *Chem. Rev.*, **1983**, *83*, 379; (f) Henry, G. D. *Tetrahedron*, **2004**, *60*, 6043.
11. (a) Weinreb, S. M. In *Comprehensive Organic Synthesis*; Trost, B. M. and Fleming, I., Ed.; Pergamon Press: Oxford, 1991; Vol. 5, pp 401. (b) Boger, D. L. In *Comprehensive Organic Synthesis*; Trost, B. M. and Fleming, I., Ed.; Pergamon Press: Oxford, 1991; Vol. 5, pp 451. (c) Kobayashi, S.; Ishitani, H. *Chem. Rev.* **1999**, *99*, 1069. (d) Jorgensen, K. A. *Angew. Chem. Int. Ed.* **2000**, *39*, 3558. (e) Carmona, D.; Lamata, M. P.; Oro, L. A. *Coord. Chem. Rev.* **2000**, *200*, 717.
12. Pégot, B.; Buu, O. N. V.; Gori, D.; Thanh, G. Vo- *Beil. J. Org. Chem.*, **2006**, 2:18.
13. Clarke, P. A.; Zaytzev, A. V.; Whitwood, A. C. *Tetrahedron Lett.*, **2007**, *48*, 5209.
14. (a) Trost, B. M. *Science*, **1991**, *254*, 1471; (b) Trost, B. M. *Angew. Chem. Int. Ed.*, **1995**, *34*, 259.
15. Bahmanyar, S.; Houk, K. N.; Martin, H. J.; List, B. *J. Am. Chem. Soc.*, **2003**, *125*, 2475.
16. Childs, G. E.; Wimonwattawattee, T.; Pooyindee, N. *Am. J. Trop. Med. Hyg.*, **1988**, *38*, 19-23.
17. Rieckmann, K.H.; Sax, L. J.; Campbell, G. H.; Mrema, J. E. *The Lancet* **1978**, 22-23
18. Mishra, R.C.; Tripathi, R.; Katiyar, D.; Tiwari, N.; Singh, D.; Tripathi, R. P. *Bioorg. Med. Chem.* **2003**, *11*, 5363-5374.